

**Exam 2016**  
**Organic Chemistry**  
**Test no.1**

Each item has only one correct answer. Choose the correct answer among options given by marking one letter:

1. Percentage of carbon for a hydrocarbon with 14.29% H is:

- a. 0%;
- b. 14.29%;
- c. 85.71%.

2. How many  $\pi$  bonds are there in acetylene?



- a. 2;
- b. 3;
- c. 1.

3. The general formula for alkynes is:

- a.  $\text{C}_n\text{H}_{2n+2}$ ;
- b.  $\text{C}_n\text{H}_{2n}$ ;
- c.  $\text{C}_n\text{H}_{2n-2}$ .

4. A solution with the hydrogen ion concentration of  $10^{-9}$  M is:

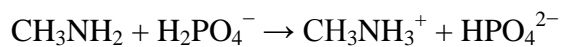
- a. neutral
- b. acidic
- c. basic

5. The correct IUPAC name for the compound  $\text{CH}_3-\text{CH}=\text{C}-\text{CH}_2-\text{CH}_3$  is:



- a. 3-methyl-3-pentene;
- b. 2-ethyl-2-butene;
- c. 3-methyl-2-pentene.

6. In the reaction:



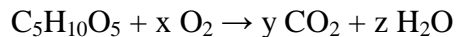
the  $\text{H}_2\text{PO}_4^-$  ion acts as:

- a. an acid;
- b. a base;
- c. a catalyst.

7. Which of the following contains only single bonds?

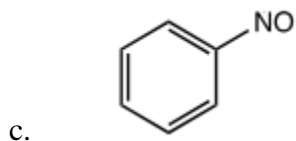
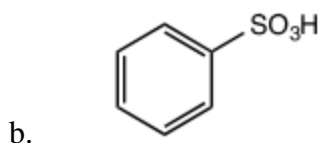
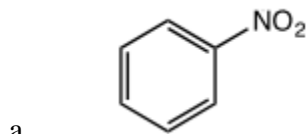
- a.  $\text{H}_2\text{CCHCH}_3$
- b.  $\text{CH}_3\text{CHO}$
- c.  $\text{CH}_3\text{NH}_2$

8. When the following equation is balanced and written with the smallest whole number coefficients, what is the coefficient of  $\text{O}_2$ ?



- a. 2
- b. 3
- c. 5

9. What is the product of the reaction of benzene with a solution of nitric acid in sulfuric acid at 50 °C?



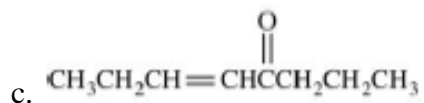
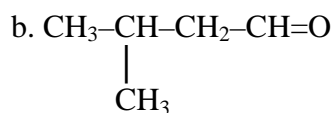
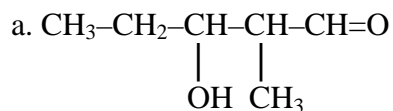
10. The amount of glucose ( $M = 180 \text{ g/mol}$ ) required to prepare 250 mL solution of 0.1 M concentration is:

- a. 180 mg;
- b. 9 g;
- c. 4.5 g.

11. Indicate which of the following biopolymers are composed of many amino acid residues:

- a. polynucleotides;
- b. polysaccharides;
- c. polypeptides.

12. Which of the following is the aldol condensation product of propanal  $\text{CH}_3\text{CH}_2\text{CHO}$ ?



Realised by,

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***Organic Chemistry***  
**Test no. 1**  
**The correct answers and scoring**

- 1. c 7.5 points
- 2. a 7.5 points
- 3. c 7.5 points
- 4. c 7.5 points
- 5. c 7.5 points
- 6. a 7.5 points
- 7. c 7.5 points
- 8. c 7.5 points
- 9. a 7.5 points
- 10. c 7.5 points
- 11. c 7.5 points
- 12. a 7.5 points

You will receive 10 points by default.

Total: 100 points